

Table 2. Geometric parameters (Å, °)

N—C1	1.510 (5)	C1—C12	1.518 (6)
N—C2	1.516 (7)	C2—C22	1.508 (5)
C1—C11	1.508 (7)	C2—C21	1.516 (5)
C1—N—C2	117.9 (3)	C22—C2—N	110.4 (5)
C11—C1—N	110.7 (3)	C22—C2—C21	112.7 (4)
C11—C1—C12	112.3 (4)	N—C2—C21	107.5 (5)
N—C1—C12	107.2 (3)		

Table 3. Hydrogen-bonding geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N—H(3)...Br ¹	0.82 (4)	2.51 (4)	3.325 (3)	173 (4)
N—H(4)...Br	0.81 (4)	2.55 (4)	3.350 (4)	171 (4)

Symmetry code: (i) $-x, y - \frac{1}{2}, 1 - z$.

The *SHELX* input files were prepared using the program *UTILITY* (Pickardt, 1994). All non-H-atoms were refined anisotropically and all H atoms were refined isotropically. The crystal was a racemic twin consisting of two individuals in the ratio 2:1, as established with a *TWIN/BASF* refinement (*SHELXL93*; Sheldrick, 1993).

Data collection: *Stadi-4 Software* (Stoe & Cie, 1995a). Cell refinement: *Stadi-4 Software*. Data reduction: *X-RED*, (Stoe & Cie, 1995b). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL93*. Molecular graphics: *PLUTON92* (Spek, 1992), *PLATON* (Spek, 1990), *ZORTEP* (Zsolnai & Pritzkow, 1994). Software used to prepare material for publication: *SHELXL93*.

This work has been supported by the Deutsche Forschungsgemeinschaft and the Volkswagen-Stiftung.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: SK1006). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Chen, C. K., Cheng, C. H. & Hseu, T. H. (1987). *Organometallics*, **6**, 868–872.
- Dell'Amico, D. B., Calderazzo, F., Giurlani, U. & Pelizzi, G. (1987). *Chem. Ber.* **120**, 955–964.
- Dell'Amico, D. B., Calderazzo, F. & Pelizzi, G. (1979). *Inorg. Chem.* **18**, 1165–1168.
- Dötz, K. H., Fischer, H., Hofmann, P., Kreißl, F. R., Schubert, U. & Weiss, K. (1983). *Transition Metal Carbene Complexes*. Weinheim: VCH.
- Fernandez, V., Moran, M., Gutierrez-Rios, M. T., Foces-Foces, C. & Cano, F. H. (1987). *Inorg. Chim. Acta*, **128**, 239–243.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Pickardt, J. (1994). *UTILITY*. Technische University Berlin, Germany.
- Prince, P., Miller, J. A., Fronczek, F. R. & Gandour, R. D. (1990). *Acta Cryst.* **C46**, 336–338.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1993). *SHELXL93. Program for Crystal Structure Refinement*. University of Göttingen, Germany.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Spek, A. L. (1992). *PLUTON92. Molecular Graphics Program*. University of Utrecht, The Netherlands.

- Stoe & Cie (1995a). *Stadi-4 Software*. Stoe & Cie, Darmstadt, Germany.
- Stoe & Cie (1995b). *X-RED. Data Reduction Program*. Stoe & Cie, Darmstadt, Germany.
- Wahlberg, A. (1978). *Acta Cryst.* **B34**, 3479–3481.
- Zsolnai, L. & Pritzkow, H. (1994). *ZORTEP. ORTEP Program for PC*. University of Heidelberg, Germany.

Acta Cryst. (1996). **C52**, 2311–2313

A Substituted *cis,trans,cis,cis*-[4.5.5]Fenestrene

JING WANG, MARC THOMMEN AND REINHART KEESE*

Institut für Organische Chemie, Universität Bern, Freiestrasse 3, CH-3012 Bern, Switzerland. E-mail: keese@ioc.unibe.ch

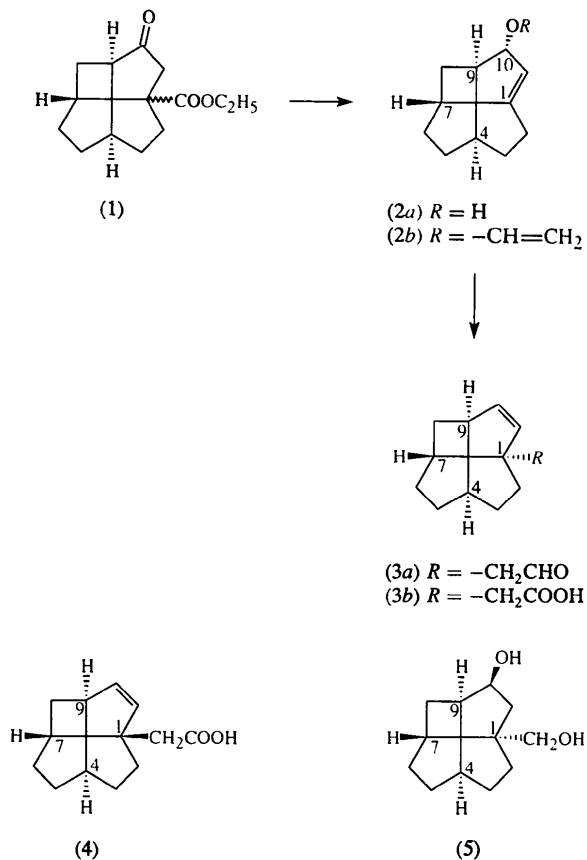
(Received 22 December 1995; accepted 18 March 1996)

Abstract

The title compound, *rel*-(1*S*,4*R*,7*R*,9*S*)-tetracyclo-[5.4.1.0^{4,12}.0^{9,12}]dodec-10-enyl-1-acetic acid, C₁₄H₁₈O₂, has been prepared and its structure elucidated. The geometry of the central C(C)₄ substructure shows a considerable distortion from an ideal tetrahedral arrangement towards planarity, with two opposite bond angles of 119.2(2) and 134.9(2)°. The other bridgehead bond angle of the *trans*-bicyclo[3.3.0]octane subunit is also large, with a value of 128.0(2)°.

Comment

Tetracyclic compounds, where four small rings share a common C atom, are of considerable interest for their propensity towards planarizing distortions at the central C atom, leading to two opposite bond angles markedly larger than 109.47° (Luef & Keese, 1992, 1993). In our systematic search for fenestranes with strong planarizing distortions, it was found that the simultaneous incorporation of *trans*-bicyclo[3.3.0]octane subunits and bridgehead substituents led to enlarged opposite bond angles at the central C atom. In order to verify this computational result and to compare the structure of (3*b*) with that of the structurally similar *cis,trans,cis,cis*-[4.5.5]fenestranediol, (5) (Hirschi, Luef, Gerber & Keese, 1992), the known compound (2*a*) was prepared from (1) by a new route, transformed *via* (2*b*) into (3*a*) by a previously published method (Grieco, Brandes, McCann & Clark, 1989) and oxidized with Jones reagent (acetone, 298 K) to compound (3*b*).



The structure of (3b) (Fig. 1) shows unique deformations of the central C(C)₄ substructure and the *trans*-bicyclo[3.3.0]octane subunit. Whereas in (4), the epimer of (3b), the bond angles of the central C atom, C4—C12—C9 and C1—C12—C7, are 124.9(2) and 122.4(2)°, respectively, the corresponding values in (3b) are 134.9(2) and 119.2(2)°. Furthermore, the bond angle C2—C1—C11 in (3b) [128.0(2)°] is considerably larger than that observed in (4) (113.0°; Grieco *et al.*, 1989). Other salient structural features are the bridgehead bond angles C3—C4—C5 and C8—C9—

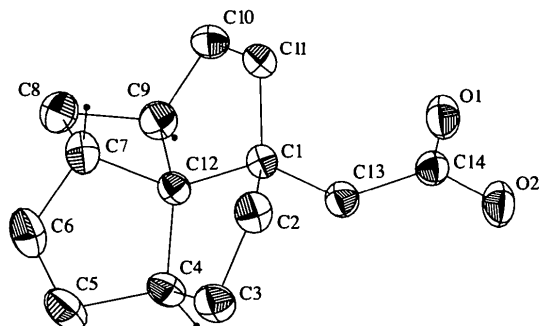


Fig. 1. The molecular structure of the title compound showing the atomic labelling. Displacement ellipsoids are shown at 30% probability. Only the three bridgehead H atoms are represented, with arbitrary radii.

C10, which are approximately 5° larger in compound (4) (117.3 and 114.8°, respectively) than in (3b) [112.3(2) and 110.0(2)°]. It is apparent that inversion of the bridgehead geometry at C1 has led to planarizing distortions at the central C atom. This is further supported by the structure of (5), where the planarizing distortions were slightly decreased due to the formal replacement of the double bond by a secondary hydroxy group. The enlargement of the bond angle C2—C1—C11 to 124.3° is also observed.

Compared with the average bond angles of 126(4) and 127(6)° determined from 19 *trans*-bicyclo[3.3.0]octane subunits retrieved from the Cambridge Structural Database (Allen & Kennard, 1993), the C4—C12—C9 bridgehead angle of 134.9(2)° in (3b) is significantly larger (Hirschi *et al.*, 1992).

Experimental

Compound (1) was reduced to the corresponding diol with LiAlH₄ in diethyl ether and reoxidized to the ketoacid with Jones reagent. Decarboxylation to the α,β -unsaturated ketone was performed with Cu(OAc)₂ and Pb(OAc)₄ in pyridine and benzene at 353 K. Subsequent reduction with 9-borabicyclo[3.3.1]nonane (9-BBN) in tetrahydrofuran gave a 1.5:1 mixture of (2a) and its 10-epimer in 45% overall yield (Guidetti-Grept, 1994). Compound (2a) was transformed *via* (2b) into (3a) by the previously published method of Grieco, Brandes, McCann & Clark (1989) and oxidized with Jones reagent (acetone, 298 K) to compound (3b). Crystals of m.p. 399–401 K were obtained from ^tBuOMe/octane solution.

Crystal data

C₁₄H₁₈O₂
M_r = 218.28
 Triclinic
 P1̄
a = 5.618(4) Å
b = 7.995(4) Å
c = 13.428(6) Å
 α = 77.85(4)°
 β = 85.55(6)°
 γ = 80.46(4)°
V = 580.9(6) Å³
Z = 2
D_x = 1.248 Mg m⁻³
D_m not measured

Mo K α radiation
 λ = 0.71073 Å
 Cell parameters from 16 reflections
 θ = 5.7–20.1°
 μ = 0.082 mm⁻¹
T = 293(2) K
 Flat elongated prism
 0.35 × 0.33 × 0.20 mm
 Colourless

Data collection

Siemens P4/RA diffractometer
 ω scans
 Absorption correction:
 empirical *via* ψ scan
 (XSCANS; Siemens, 1994)
T_{min} = 0.939, *T_{max}* = 1.000
 3129 measured reflections
 3129 independent reflections

2222 observed reflections
 [*I* > 2 σ (*I*)]
 θ_{max} = 24.99°
h = -1 → 7
k = -11 → 11
l = -18 → 18
 3 standard reflections
 monitored every 100 reflections
 intensity decay: 12%

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\max} < 0.001$
$R(F) = 0.0419$	$\Delta\rho_{\max} = 0.149 \text{ e } \text{\AA}^{-3}$
$wR(F^2) = 0.1181$	$\Delta\rho_{\min} = -0.165 \text{ e } \text{\AA}^{-3}$
$S = 1.029$	Extinction correction: none
3103 reflections	Atomic scattering factors
218 parameters	from <i>International Tables</i>
H atoms freely refined	for <i>Crystallography</i> (1992),
isotropically	Vol. C, Tables 4.2.6.8 and
$w = 1/[\sigma^2(F_o^2) + (0.0518P)^2$	6.1.1.4)
$+ 0.2214P]$	
where $P = (F_o^2 + 2F_c^2)/3$	

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)
$$U_{eq} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i \cdot a_j$$

	x	y	z	U_{eq}
O1	0.5489 (3)	0.6584 (2)	0.40420 (11)	0.0626 (5)
O2	0.1972 (3)	0.6311 (2)	0.48825 (12)	0.0687 (5)
C1	0.3640 (3)	0.9838 (2)	0.27865 (13)	0.0404 (5)
C2	0.5198 (4)	1.0752 (3)	0.3318 (2)	0.0545 (6)
C3	0.3332 (5)	1.2246 (3)	0.3615 (2)	0.0614 (6)
C4	0.1182 (4)	1.2628 (2)	0.2886 (2)	0.0515 (5)
C5	0.0933 (6)	1.4462 (3)	0.2231 (2)	0.0729 (8)
C6	0.3092 (6)	1.4409 (3)	0.1478 (2)	0.0755 (8)
C7	0.3375 (5)	1.2613 (3)	0.1242 (2)	0.0577 (6)
C8	0.1534 (5)	1.2307 (3)	0.0518 (2)	0.0685 (7)
C9	0.0960 (4)	1.0732 (3)	0.13260 (15)	0.0532 (6)
C10	0.2846 (5)	0.9173 (3)	0.1249 (2)	0.0607 (6)
C11	0.4427 (4)	0.8777 (3)	0.1976 (2)	0.0535 (6)
C12	0.2110 (3)	1.1405 (2)	0.21572 (13)	0.0415 (5)
C13	0.2047 (4)	0.8798 (3)	0.3597 (2)	0.0478 (5)
C14	0.3374 (4)	0.7121 (2)	0.41852 (14)	0.0454 (5)

Table 2. Selected geometric parameters (\AA , $^\circ$)

O1—C14	1.209 (2)	C5—C6	1.519 (4)
O2—C14	1.310 (2)	C6—C7	1.516 (3)
C1—C11	1.513 (3)	C7—C8	1.554 (3)
C1—C12	1.527 (3)	C7—C12	1.594 (3)
C1—C2	1.529 (3)	C8—C9	1.540 (3)
C1—C13	1.546 (3)	C9—C10	1.512 (3)
C2—C3	1.550 (3)	C9—C12	1.560 (3)
C3—C4	1.564 (3)	C10—C11	1.323 (3)
C4—C12	1.529 (3)	C13—C14	1.514 (3)
C4—C5	1.533 (3)		
C11—C1—C12	100.02 (15)	C1—C12—C4	106.21 (15)
C11—C1—C2	128.0 (2)	C1—C12—C9	106.72 (15)
C12—C1—C2	100.1 (2)	C4—C12—C9	134.9 (2)
C11—C1—C13	107.4 (2)	C1—C12—C7	119.2 (2)
C12—C1—C13	110.8 (2)	C4—C12—C7	103.0 (2)
C2—C1—C13	109.2 (2)	C9—C12—C7	86.78 (15)

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: PEANUT (Hummel, Hauser & Bürgi, 1990). Software used to prepare material for publication: SHELXL93.

This work has been generously supported by the Swiss National Science Foundation and the Stipendiefonds der Basler Chemischen Industrie. The authors are grateful to Dr C. Monnier, Ciba, Marly, for the generous supply of intermediates and thank M. Förtsch for the X-ray measurements.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry and torsion angles have been deposited with the IUCr (Reference: CF1078). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Allen, F. H. & Kennard, O. (1993). *Chem. Des. Autom. News*, **8**, 31–37.
- Grieco, P. A., Brandes, E. B., McCann, S. & Clark, J. D. (1989). *J. Org. Chem.* **54**, 5849–5851.
- Guidetti-Grept, R. (1994). PhD thesis, University of Bern, Switzerland.
- Hirschi, D., Luef, W., Gerber, P. & Keese, R. (1992). *Helv. Chim. Acta*, **75**, 1897–1908.
- Hummel, W., Hauser, J. & Bürgi, H.-B. (1990). *J. Mol. Graph.* **8**, 214–220.
- Luef, W. & Keese, R. (1992). *J. Mol. Struct. (Theochem.)*, **257**, 353–368.
- Luef, W. & Keese, R. (1993). *Advances in Strain in Organic Chemistry*, Vol. 3, edited by B. Halton, pp. 229–267. Greenwich, Ct: JAI Press Ltd.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siemens (1994). *XSCANS Users Manual*. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Acta Cryst. (1996). **C52**, 2313–2315

3a,7a-Dibromo-2-butyl-4,7-methano-1,3,3a,4,7,7a-hexahydro-2H-isoindole-1,3-dione from the Reaction of 3,4-Dibromo-1-butylmaleimide with Cyclopentadiene

ROY L. BEDDOES, YUEKUN ZHAO AND JOHN A. JOULE*

Chemistry Department, The University of Manchester, Manchester M13 9PL, UK. E-mail: j.a.joule@man.ac.uk

(Received 27 March 1996; accepted 14 June 1996)

Abstract

It is shown that reaction of 1-butylpyrrole successively with *N*-bromosuccinimide then nitric acid gives 3,4-dibromo-1-butylmaleimide since the latter then undergoes Diels–Alder cycloaddition with cyclopentadiene to give the title compound, $C_{13}H_{15}Br_2NO_2$.

Comment

Interest in the cycloaddition reactions of oxidopyraziniums (Yates, Peters, Beddoes, Scopes, & Joule, 1995) drew our attention to the recent report (Brockmann